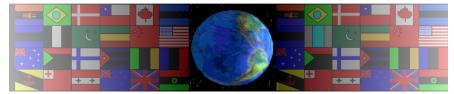
REVISION CARDS - METAL FINISHES - 6

ELECTROPLATING

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On behalf of The World Association of Technology Teachers

W.A.T.T.



World Association of Technology Teachers

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REVISION CARDS - METAL FINISHES - 6

ELECTROPLATING - 2

ELECTROPLATING WITH COPPER

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IONIC SOLUTION

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ELECTROPLATING - 3 The copper anode is ionised once the SILVER PLATED CHROME ELECTROPLATED GOBLET electrical current is allowed to flow. The TAP ionic solution allows the positively charged copper atoms to flow to the negatively charged base metal, where they are deposited on the surface. This produces an effective coating of copper. GOLD PLATED COPPER PLATED SILVER PLATED WATCH MUG TEAPOT NEGATIVE CATHODE www.technologystudent.com © 2017

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ELECTROPLATING - 1

In industry, electroplating is the process whereby a cheap base metal, is coated with a much more expensive metal. in order to make it visually attractive and aesthetically pleasing (gold and silver plating are examples). Electroplating is usually a decorative process and is often used to increase the visual appeal of cheaper jewellery. It also serves to provide the surface with a level of protection against corrosion. Some everyday products such as bathroom taps have been electroplated with chrome for decoration, as well as corrosion resistance. Electroplating is also used to apply a conductive surface to metals, that are of low conductivity or non-conductive.

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1. What is electroplating? (include the reasons for electroplating) 5 marks

2. List six products / components that have been electroplated? 4 marks